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**Basic Stoichiometry Post Lab Homework Exercises Answers [PDF]**

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stoichiometry. When the reactants are present in the correct amounts, the reaction will produce products. When the reactants are present in the correct amounts, the reaction will produce products.

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stoichiometry When the reactants are present in the correct amounts, the reaction will produce products. What happens if there are more or less of some of the reactants present?

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Post-lab Discussion The Ag/AgNO<sub>3</sub> ratio in either reaction scenario would theoretically be 1. ... complete the lab report. 3. Worksheet 2: Stoichiometry and Percent Yield

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Stoichiometry I. Introduction/ Purpose: Stoichiometry is the study of the quantitative, or measurable, relationships that exist in chemical formulas and also chemical reactions. The calculations of a stoichiometry problem depend upon balanced chemical equations.

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